

Py GC-MS analysis of organic binders in Cultural Heritage



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NPI Conference, Prague 9 November 2022

Motivation for the analysis of art materials



Art material analysis important in order to:

- Better understand the **Master's technique**, her/his artistic, socio-cultural, economic background, and the resources and materials available to him
- **Identify any alterations**, modifications or later conservative intervention
- **Assess the authenticity** of a particular piece of art
- **Reveal potential changes** of the materials used by the artist over time, or under the influence of light and environmental conditions

Organic materials in art objects

Different binder reveal different painting technique

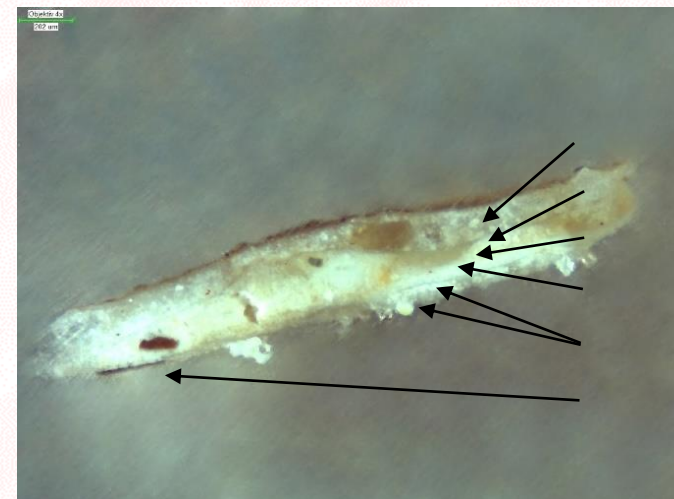
- **Encaustic painting:** wax
- **Oil paint:** siccative oil (mainly linseed oil)
- **Tempera painting:** based on a proteinaceous component (egg, gelatin, animal glue, casein, or mixture of these) lipid additives, as siccative oils (if present in higher amounts the technique is called tempera grassa)
- **Aquarelle painting:** mainly arabic gum
- **Modern painting:** synthetic resins



Challenges in the analysis of organic paint constituents

Critical consideration of the following factors necessary when planning an analytical investigation:

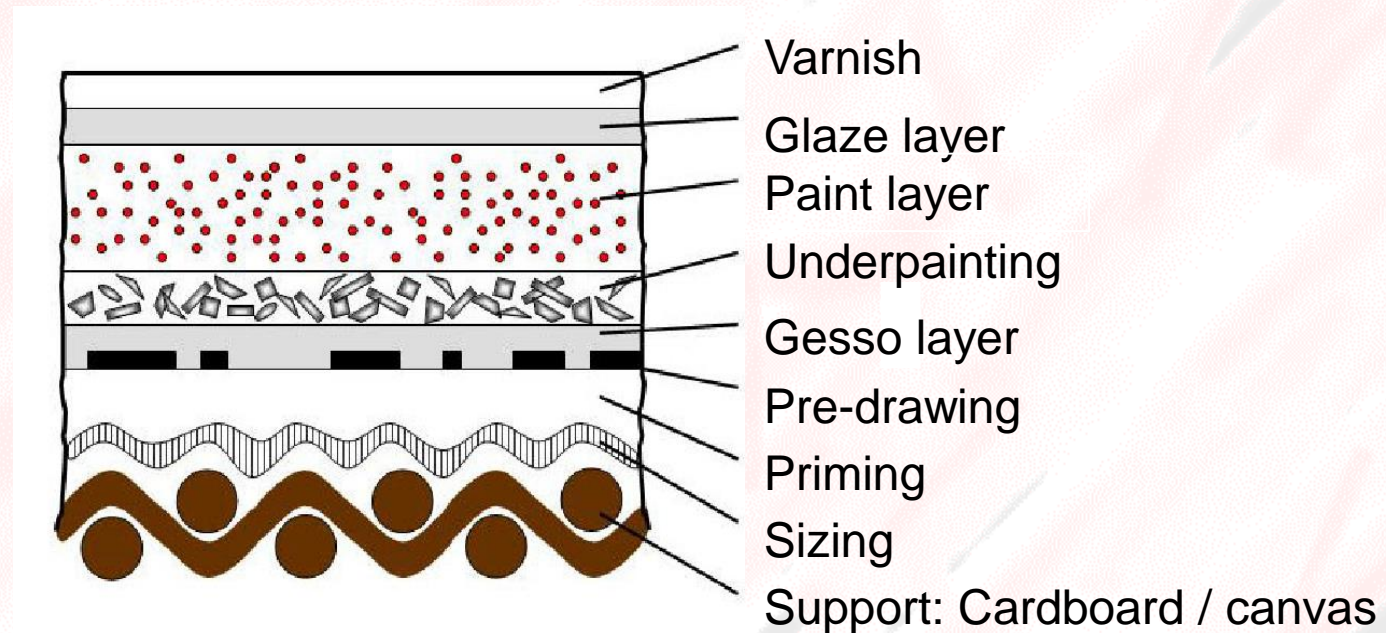
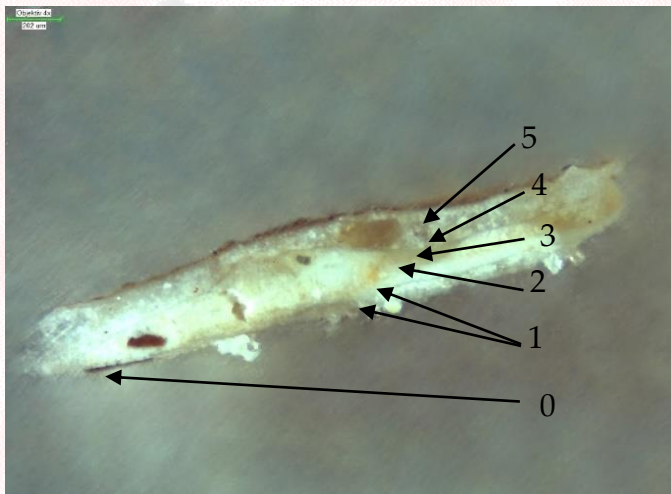
- **Several** organic natural and synthetic **substances** are often simultaneously present in the layered structure;
- **Non-original compounds**, formed as a result of aging or introduced by restoration treatments and pollution, are generally also present;
- A **very low amount of organic matter** (a few percent in the overall weight or even lower) is generally encountered in a minute heterogeneous paint sample ($\ll 1$ mg)



Paint sample under the microscope 50x

Organic materials in art objects

- **Binders** if used as a matrix of the paint
- **Varnishes** if applied as a protective layer
- **Consolidants** if used for the conservation of a fragile paint (⇒ can complicate the understanding of the original painting technique !)



Organic materials used in artworks



Category	Organic materials	Uses
Proteins	Egg, milk and casein, animal glue, silk, wool, vegetable proteins (e.g. garlic, beans), human and animal tissues (e.g. mummies)	Paint binders, adhesives, textiles, commodities, parchment
Glycerolipids	Animal fats, vegetable oils (e.g. palm oil, olive oil) including drying oils (e.g. linseed, walnut, poppy seed)	Paint binders, varnishes, illuminants, commodities, ingredients of cosmetic and pharmaceutical preparations
Waxes	Beeswax, spermaceti, Chinese wax, lanolin (animal waxes); carnauba, candelilla, Japan wax, esparto (vegetable waxes); paraffin, ozokerite (fossil waxes)	Paint binders, coatings, sealants, writing tablets, ingredients of cosmetic and pharmaceutical preparations, sculptures
Natural resins	Pine resins, sandarac, copals, mastic, dammar, amber, frankincense, benzoe, styrax, myrrh, (plant resins); shellac (animal resin); tar and pitch (from thermal treatment of plant resins or wood)	Varnishes, coatings, waterproofing materials, paint binders, ingredients of cosmetic and pharmaceutical preparations
Polysaccharide materials	Starch, cellulose, plant gums (arabic gum, tragacanth, karaya, ghatti, guar, locust bean, fruit tree gum)	Paper, paint binders, adhesives
Bituminous materials	Bitumen, asphalt	Moulding materials, adhesive, pigment
Organic dyes	Cochineal, madder, kemes, saffron, purple, indigo, synthetic dyes	Colourants for dyeing textiles, paint materials
Synthetic polymers	Polyacrylates, cellulose nitrate, phenolic resins, polyethylene, poly(vinyl acetate), polystyrene	Paint binders, varnishes, coatings, consolidants, sculptures

these materials are characterized by a macromolecular nature: in some cases they are natural polymers, others undergo oligomerization or cross-linking reactions as an effect of exposure to light and air . Due to their macromolecular nature, analytical pyrolysis is a fast and efficient approach for identifying such organic materials in samples from works of art.

Natural polymers

Undergo polymerization

Materials and Methods

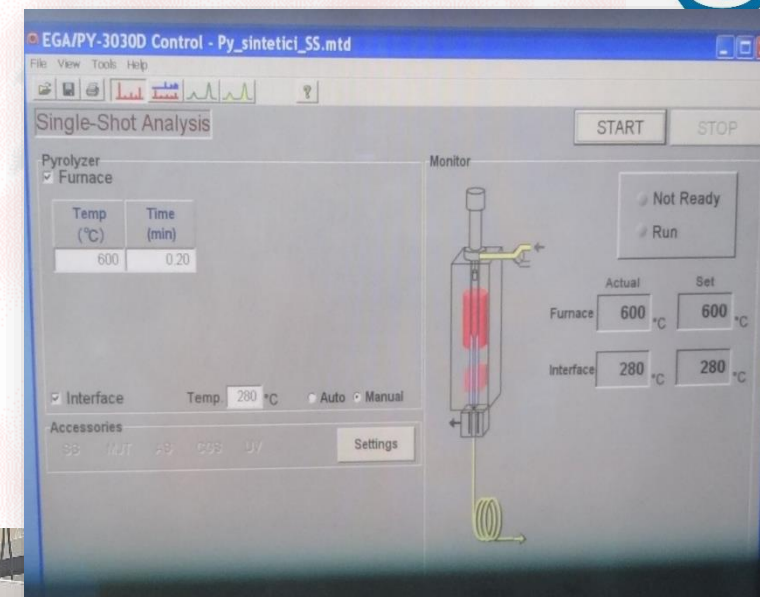
- The instrumentation consisted of a micro-furnace Multi-Shot Pyrolyzer EGA/Py-3030D (Frontier Lab, Japan) connected to an Agilent chromatograph 8890 and a single quadrupole MSD Agilent 5977B

This technique was selected due to its minimum sample size and pre-treatment requirements



Materials and Methods

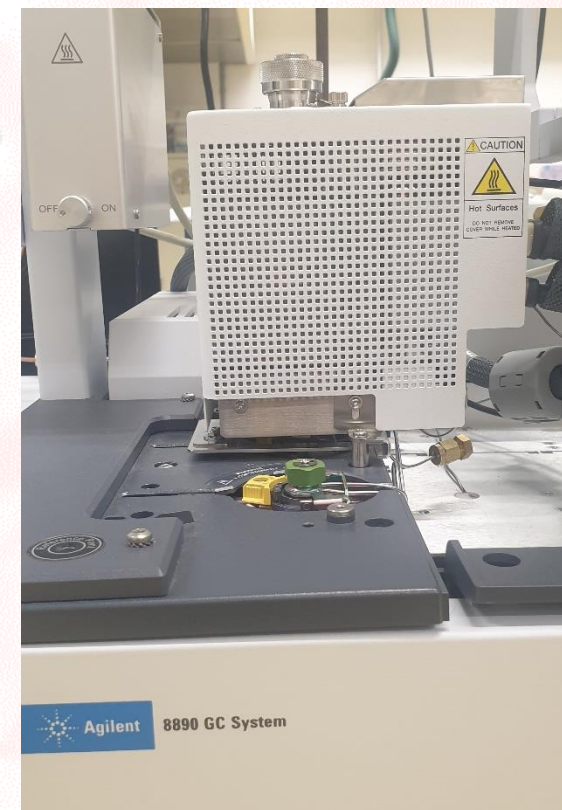
- Single shot pyrolysis was used, pyrolysis conditions were optimized as follows: pyrolysis chamber temperature 550 °C, time: 1 min, T interface 280 °C. The GC injector Temperature was 280 °C. The GC injection port was operated in split mode with a split ratio of 1:10
- MSD ChemStation (Agilent Technologies) software was used for data analysis, and peak assignment was based on a comparison with mass spectra libraries (NIST)



Sample preparation and Pyrolysis Conditions

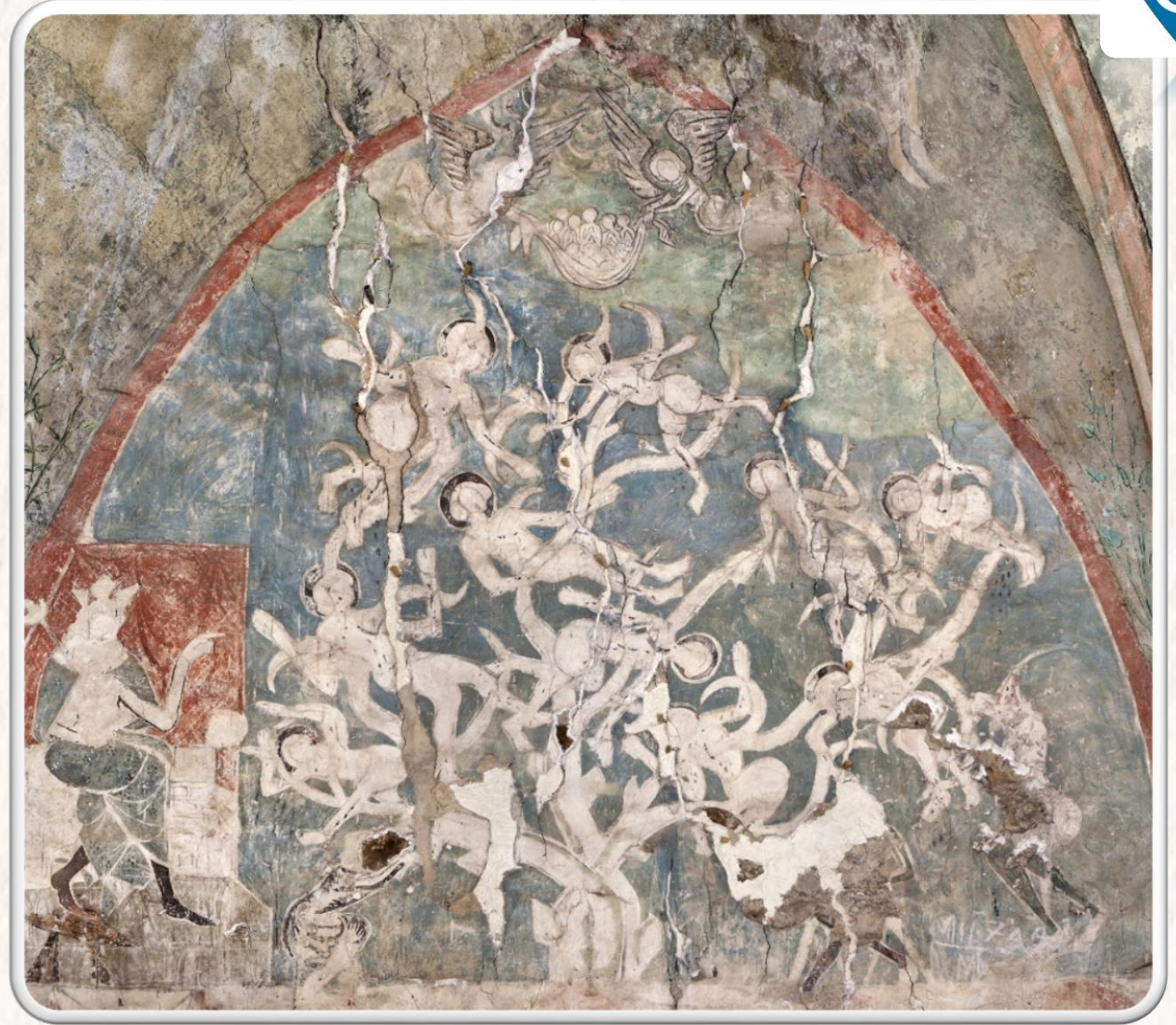
The samples (~0.2 mg) were placed in platinum sample cups and then derivatized with 5 μ l of HMDS.

The cups were placed on top of the pyrolyzer at ambient temperature and then placed into the furnace.



Py/GC-MS Application

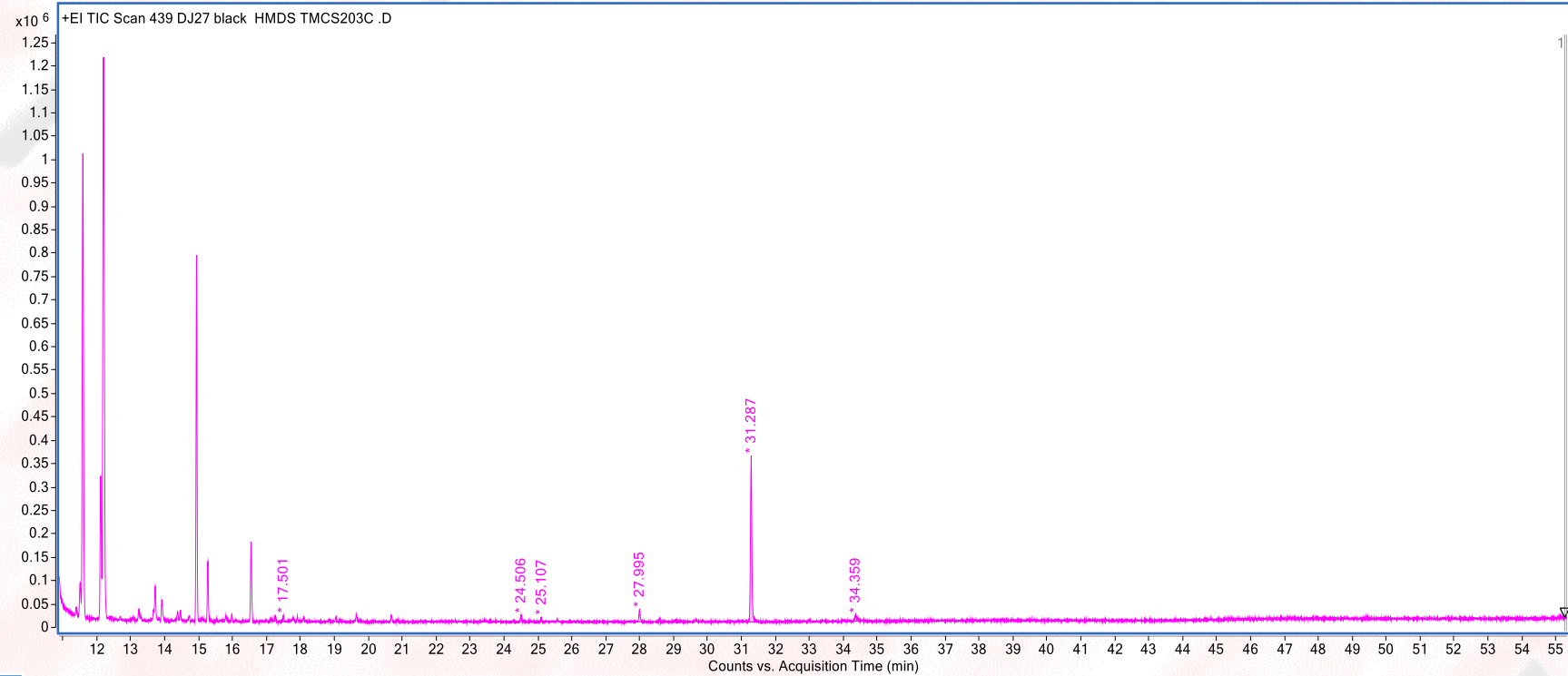
Case Studies
Wall
painting
16th Century



Sample 1



- Picture of the sample under microscope 50x



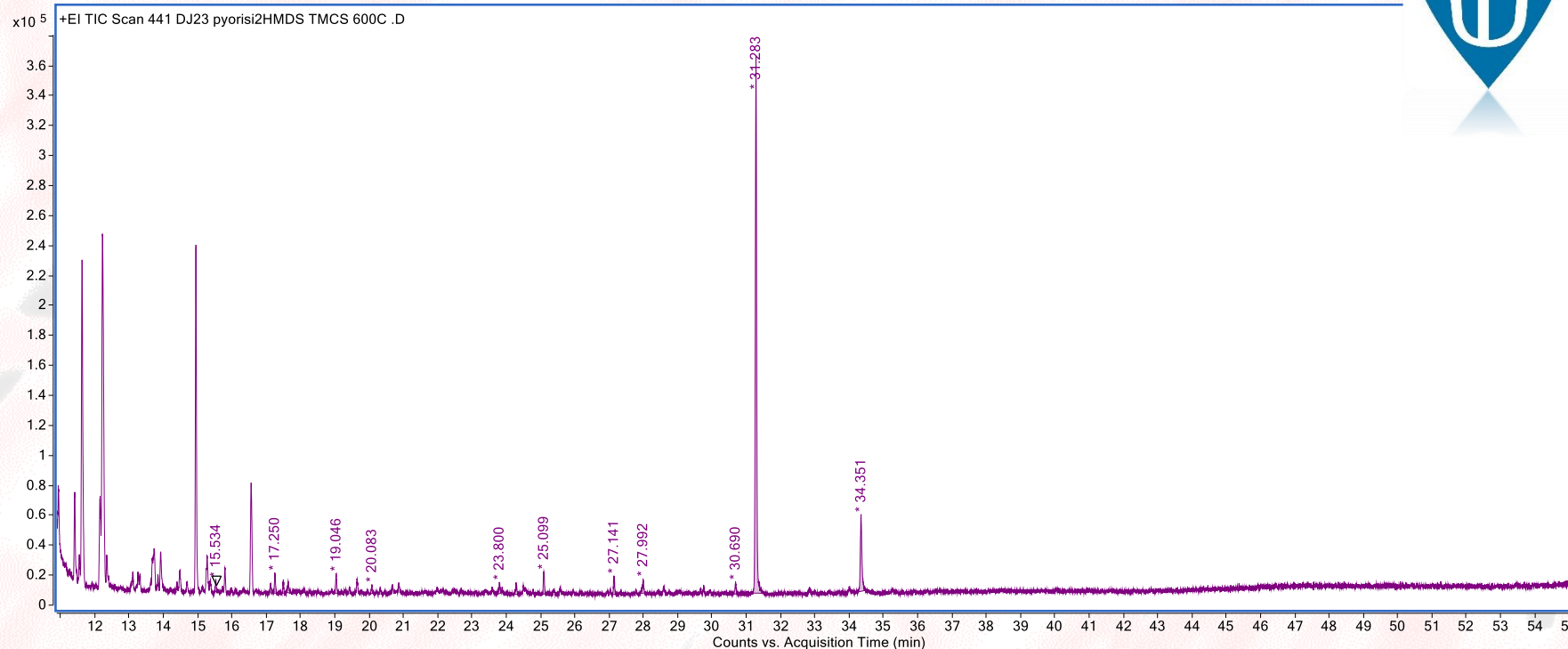
No	Ret Time	Compound
3	17.5	Glycerol
8	24.5	Octadecanoic acid
10	27.9	Tetradecanoic acid
11	31.3	Hexadecanoic acid
12	34.3	Stearic acid

Proteinaceous tempera

Sample 2



- Picture of the sample under microscope 50x



No	Ret Time	Compound
1	15.5	heptanoic acid ■
2	17.2	octanoic acid ■
3	17.5	glycerol ▲
4	17.6	silanol, trimethyl-, phosphate (3:1) ▲
5	19	nonanoic acid ■
6	23.8	Silane, [1,2,3-benzenetriyltris(oxy)]tris[trimethyl-
7	24.5	Octadecanoic acid ■
8	27.1	Azelaic acid ■
9	27.9	Tetradecanoic acid ■
10	31.3	Hexadecanoic acid ■
11	34.3	Stearic acid ■

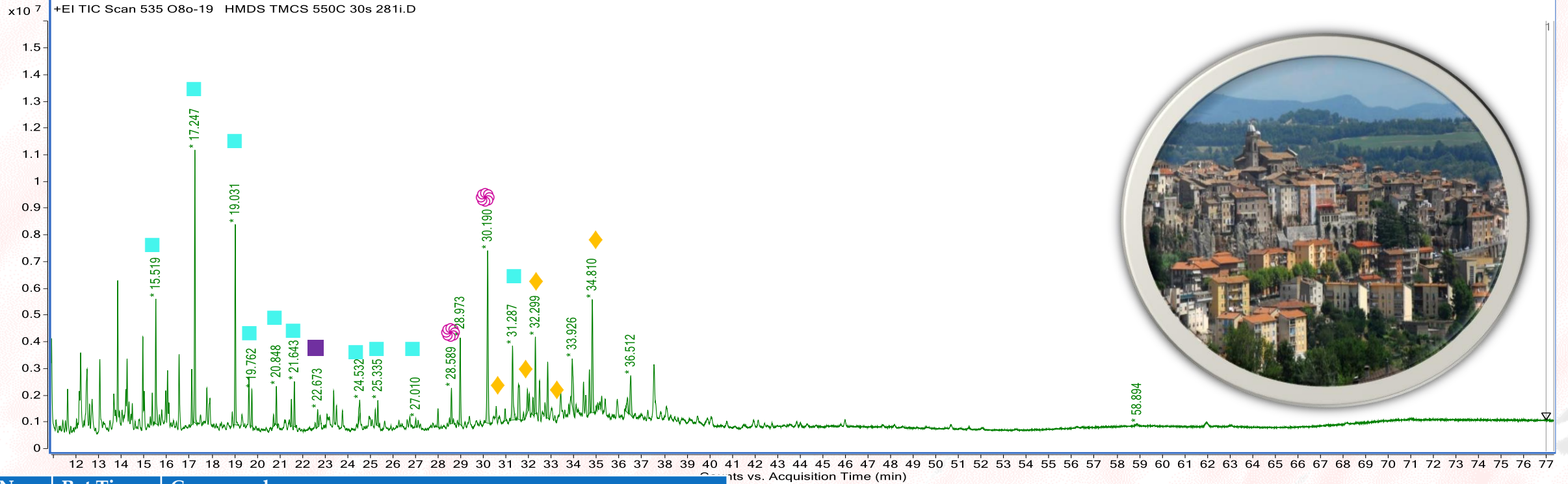


Py/GC-MS Application

Orte polychrome
objects



Case study 2: Orte Polychrome objects



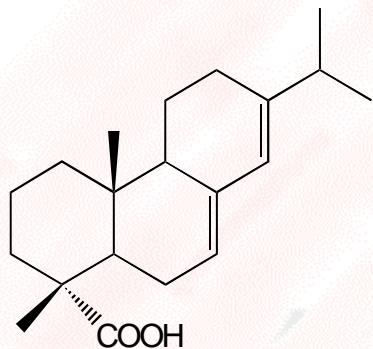
No	Ret Time	Compound
1	15.5	heptanoic acid
2	17.2	octanoic acid
3	19.0	nonanoic acid
4	20.8	decanoic acid
5	21.69	pentadecane
6	22.6	undecenoic acid
7	24.5	dodecanoic acid
8	25.3	octanedioic acid
9	27.1	nonadioic acid (azelaic)
10	28.5	phtalic acid deriv

No	Ret Time	Compound
11	30.2	dibutyl phthalate
12	30.9	dehydroabiatic acid
13	31.3	hexadecanoic acid (palmitic)
14	31.5	pimaric acid
15	32.2	phenanthrene,2,7-dimethyl
16	32.8	10,18-Bisnorabieta-5,7,9(10),11,13-pentaene
17	34.8	phenanthrene, 1-methyl-7-(1-methylethyl)-

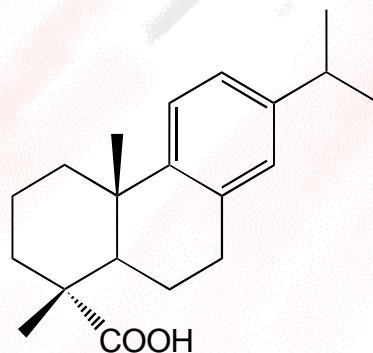
Diterpenic molecules in resins from conifer plants order



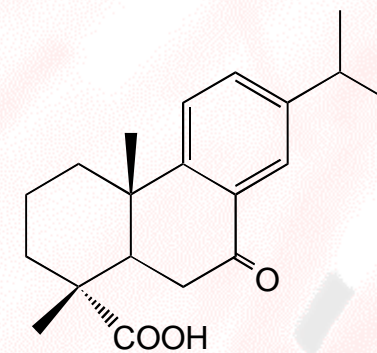
Abietane skeleton



Abietic acid

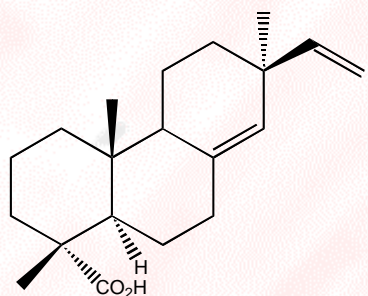


Dehydroabietic acid

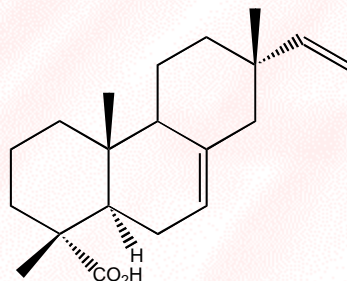


7-oxodehydroabietic acid

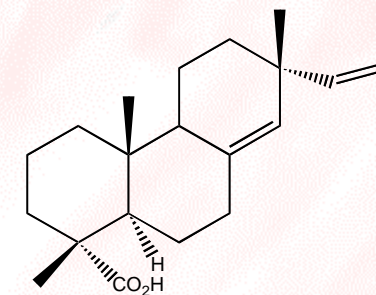
Pimarane skeleton



Pimaric acid



Isopimaric acid



Sandaracopimaric acid

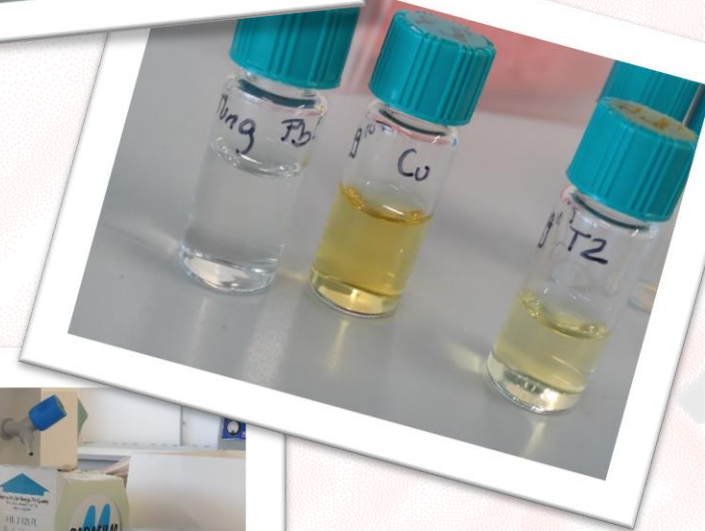
Conclusions

Py-GC/MS is rapid, avoids sample wet chemical workup, avoiding sample loss and contamination, and has a low sample requirement.

It allows the determination, in a single step, of polymeric materials and low molecular weight components

Pyrolysis is a relatively inexpensive technique, especially if compared with the classical wet analytical procedures that are required prior to GC/MS analyses.

Py-GC/MS technique overcomes the major disadvantages encountered in GC/MS, i.e. undetectability of synthetic polymers and unpredictable interferences

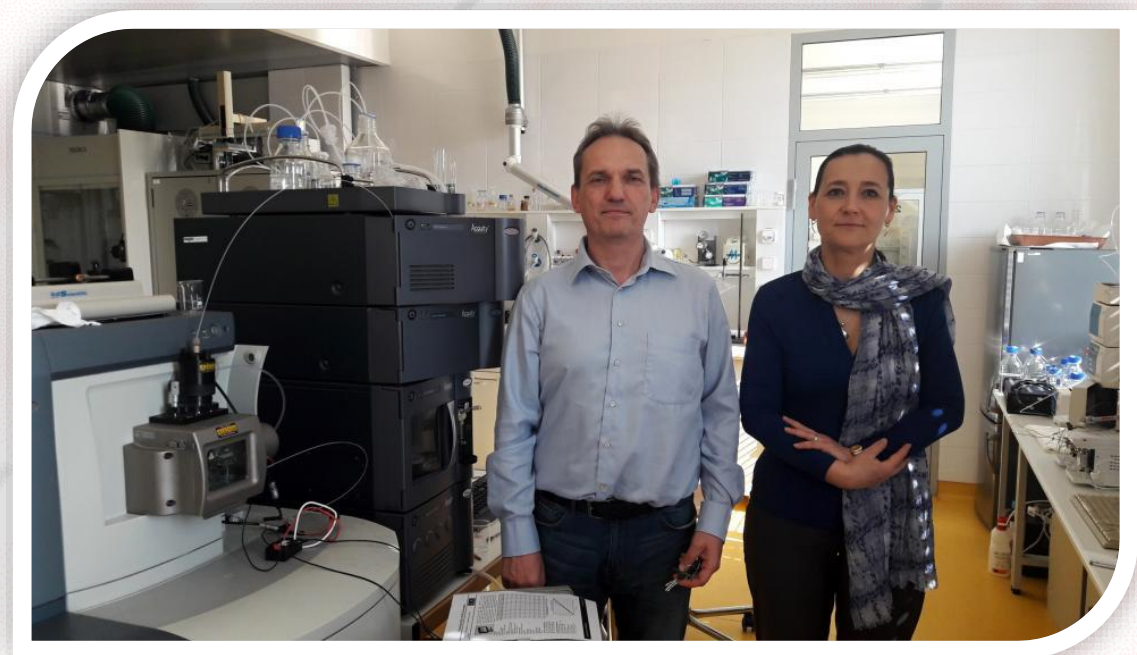


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Special thanks to:

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Thank you for your attention!

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